

Synthesis of Pyrrolo-isoquinolines Related to the Lamellarins Using Silver-Catalyzed Cycloisomerization/Dipolar Cycloaddition

Shun Su and John A. Porco, Jr.*

Department of Chemistry and Center for Chemical Methodology and Library Development (CMLD-BU), Boston University, Boston, Massachusetts 02215

Received April 19, 2007; E-mail: porco@bu.edu

The lamellarin alkaloids are a family of novel marine natural products containing a highly substituted pyrrolo-isoquinoline core.¹ Lamellarin D **1** (Figure 1a) is a potent inhibitor of human topoisomerase II² and was recently shown to act on mitochondria to induce apoptosis.³ Another member of the family, lamellarin α -20-sulfate **2**, displays selective inhibition against HIV-1 integrase in vitro.^{4,5} We have considered development of methodology for rapid access to angular, pyrrolo-isoquinoline core structures⁶ related to the lamellarins⁷ using variants of the cycloisomerization of alkynyl imines⁸ involving dipolar cycloaddition of derived azomethine ylides.⁹ In this approach, treatment of alkynyl imines **3** with alkynophilic metal catalysts should afford azomethine ylides **4** which may undergo dipolar cycloaddition in the presence of dipolarophiles. Subsequent aromatization may lead to the formation of pyrrole frameworks **5** (Figure 1b).

To identify suitable conditions for the proposed metal-catalyzed domino cycloisomerization/dipolar-cycloaddition process, reaction screening¹⁰ involving alkynyl *N*-benzylidene glycinate **6**, dimethyl acetylene dicarboxylate (DMAD) **7**, and a series of alkynophilic metals was carried out in DCE using DIEA as base (50 °C).¹¹ Fortunately, a number of catalysts including the Au, Ag, and Cu salts shown in Figure 2 and Pd(OAc)₂ were able to facilitate the transformation. AgOTf and AgSbF₆ provided the best results in comparison to other metals investigated and were effective in catalytic quantities (10 mol %). Control experiments in the absence of metal salts led to decomposition ruling out the possibility of spontaneous formation of **8** under thermal conditions.¹²

To improve the efficiency of the cycloisomerization/dipolar-cycloaddition process, further optimization was carried out using imine **6**, DMAD **7**, and AgOTf (10 mol %). Parameters including solvents, bases, reaction temperature, and inclusion of external oxidants (e.g., DDQ, MnO₂) were investigated. Reactions conducted at 60 °C under ambient conditions in toluene using 2,6-di-*tert*-butyl-4-methylpyridine (DTBMP) as base in the absence of oxidants afforded optimal results. Using this condition, select dipolarophiles were reacted with a number of alkynyl imines to generate diverse pyrrolo-isoquinolinone structures **8** and **24–37** (Table 1). Silver-mediated pyrrolo-isoquinoline formation was found to be workable with ortho alkynyl imines **9–17** bearing aryl, cyclopropane, propargyl ether, trimethylsilyl, terminal alkyne, and enyne functionality, as well as substrates with electron withdrawing and donating substituents on the aromatic backbone (entries **2**, **5**, and **7–9**). In addition to DMAD, alkynes **19–23** were also investigated to provide lamellarin-type structures with unsymmetrical substituents on the pyrrolo-isoquinoline core. Reactions involving alkynes **19–23** typically afforded the desired products with lower yield compared to those using DMAD owing to diminished reactivity. The addition of Lewis acids¹³ to further activate the dipolarophiles led to only marginal improvement in reaction yields. Cycloisomerization/dipolar-cycloaddition was generally found to be highly

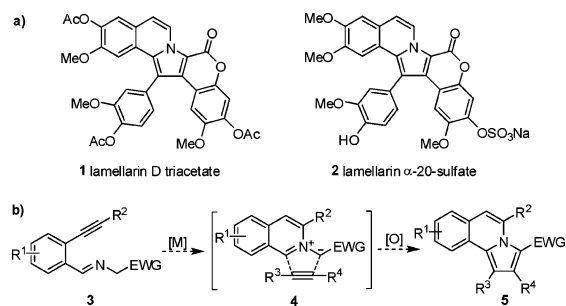


Figure 1.

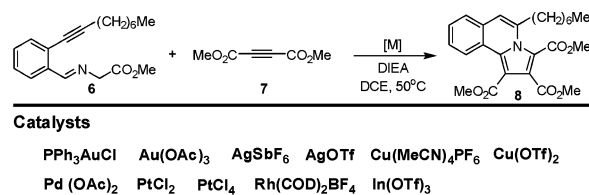


Figure 2. Reaction screening.

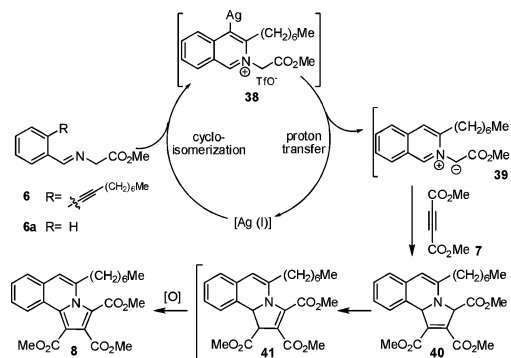


Figure 3. Proposed reaction pathway.

regioselective, providing **33–34** and **36–37** as single regioisomers.¹¹ However, reaction involving alkyne **21** produced **35** as a 1.3:1 mixture of regioisomers¹⁴ and a desilylation product derived from **37** was observed when using alkyne **23** as dipolarophile. Thus far, alkynes lacking electron withdrawing groups have not been workable as dipolarophiles.

To probe possible reaction pathways for the cycloisomerization/dipolar-cycloaddition, we conducted a series of NMR experiments. Treatment of substrate **6** with AgOTf, DMAD, and DTBMP (toluene-*d*₈, room temperature, 6 min) afforded dihydropyrrole **41**, presumably via dihydropyrrole **40** (Figure 3). The structure of intermediate **41** was determined by extensive 2-D NMR experiments.^{11,15} Subsequent thermolysis of a solution of **41** at 60 °C under aerobic conditions led to the formation of **8**. Additional experiments were conducted to rule out a dipolar-cycloaddition/

Table 1. Reaction Scope^a

entry	R	R ¹ , R ²	R ³ , R ⁴	product	yield(%) ^{a,b}
1		H, H	MeO ₂ C, CO ₂ Me		72
2		H, F	MeO ₂ C, CO ₂ Me		59
3		H, H	EtO ₂ C, CO ₂ Et		83
4		H, H	MeO ₂ C, CO ₂ Me		78
5		OMe, OMe	MeO ₂ C, CO ₂ Me		78
6		H, H	MeO ₂ C, CO ₂ Me		63
7		OCH ₂ O	MeO ₂ C, CO ₂ Me		68
8		OMe, OMe	MeO ₂ C, CO ₂ Me		70
9		OMe, OMe	MeO ₂ C, CO ₂ Me		42
10		H, H	MeO ₂ C, CO ₂ Me		29
11		H, H	OHC, C ₆ H ₅		46
12		H, H	MeOC, Ph		56
13		H, H	MeO ₂ C, Ph		57 ^c
14		H, H	MeO ₂ C, H		52
15		H, H	EtO ₂ C, SiMe ₃		28 ^{c,d}

^a Yields based on the corresponding alkynyl benzylaldehydes.¹¹ ^b R⁵ = Me, for entries 7 and 10, R⁵ = 'Bu. ^c Yb(OTf)₃ (5 mol %) was added. ^d 16% of a desilylation product was also isolated.

cycloisomerization pathway. Reaction of *des*-alkyne substrate **6a** under conditions employed for formation of **8** led to recovered starting material. In the absence of DMAD, treatment of **6** with AgOTf and DTBMP (toluene-*d*₈, room temperature) led to the formation of a complex mixture by ¹H NMR analysis. Mass spectrograph analysis¹¹ of the mixture indicated the presence of **39** and the corresponding dimer.^{11,16} Subsequent addition of DMAD

also led to the formation of **41**. The available mechanistic data supports initial cycloisomerization of **6** to isoquinolinium species **38**.^{8c} Subsequent proton transfer and regeneration of Ag (I) affords azomethine ylide **39**, which may be followed by dipolar cycloaddition to dihydropyrrole **40**. Isomerization and final oxidation affords pyrrolo-isoquinoline **8**.

In summary, we have developed an efficient synthesis of pyrrolo-isoquinolines related to the lamellarin natural products involving domino cycloisomerization/dipolar-cycloaddition of readily available alkynyl *N*-benzylidene glycinate. Mechanistic studies revealed Ag(I)-catalyzed cycloisomerization to an azomethine ylide as a key step for formation of pyrrolo-isoquinoline structures. Further studies, including applications toward the syntheses of the lamellarin alkaloids, are currently in progress and will be reported in due course.

Acknowledgment. Financial support from the NIH-NIGMS CMLD initiative (Grant P50 GM067041) and Merck Research Laboratories is gratefully acknowledged. We thank Dr. Emil Lobkovsky (Cornell University) for X-ray crystal structure analyses and Dr. Aaron B. Beeler and Ms. Ji Qi (Boston University) for experimental assistance.

Supporting Information Available: Experimental procedures and characterization data for all new compounds. X-ray crystal structure coordinates and files in CIF format. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- Bailly, C. *Curr. Med. Chem. Anti-Cancer Agents* **2004**, *4*, 363.
- Marco, E.; Laine, W.; Tardy, C.; Lansiaux, A.; Iwao, M.; Ishibashi, F.; Bailly, C.; Gago, F. *J. Med. Chem.* **2005**, *48*, 3796.
- Kluz, J.; Gallego, M.-A.; Loyens, A.; Beauvillain, J.-C.; Fernandez, Sousa-Faro, J.-M.; Cuevas, C.; Marchetti, P.; Bailly, C. *Cancer Res.* **2006**, *66*, 3177.
- Reddy, M. V. R.; Rao, M. R.; Rhodes, D.; Hansen, M. S. T.; Rubins, K.; Bushman, F. D.; Venkateswarlu, Y.; Faulkner, D. J. *J. Med. Chem.* **1999**, *42*, 1901.
- Aubry, A.; Pan, X.-S.; Fisher, L. M.; Jarlier, V.; Cambau, E. *Antimicro. Agents Chemother.* **2004**, *48*, 1281.
- For the synthesis of pyrrolo-isoquinolines, see: (a) Seregin, I. V.; Gevorgyan, V. *J. Am. Chem. Soc.* **2006**, *128*, 12050. (b) Yavari, I.; Sabbaghian, M.; Hossaini, Z. *Syn. Lett.* **2006**, 2501. (c) Kobayashi, M.; Tanabe, M.; Kondo, K.; Aoyama, T. *Tetrahedron Lett.* **2006**, *47*, 1469.
- Synthetic studies: (a) Banwell, M.; Flynn, B.; Hockless, D. *Chem. Commun.* **1997**, 2259. (b) Handy, S. T.; Zhang, Y.; Bregman, H. J. *Org. Chem.* **2004**, *69*, 2346. (c) Handy, S. T.; Zhang, Y. *Org. Prep. Proc. Int.* **2005**, *37*, 411. (d) Pla, D.; Marchal, A.; Olsen, C. A.; Albericio, F.; Alvarez, M. J. *Org. Chem.* **2005**, *70*, 8231. (e) Ploypradith, P.; Petchmanee, T.; Sahakitpichan, P.; Litvinas, N. D.; Ruchirawat, S. *J. Org. Chem.* **2006**, *71*, 9440.
- (a) Huang, Q.; Larock, R. C. *J. Org. Chem.* **2003**, *68*, 980. (b) Dai, G.; Larock, R. C. *J. Org. Chem.* **2003**, *68*, 920. (c) Rubin, M.; Sromek, A. W.; Gevorgyan, V. *Syn. Lett.* **2003**, 2265. (d) Ohtaka, M.; Nakamura, H.; Yamamoto, Y. *Tetrahedron Lett.* **2004**, *45*, 7339. (e) Asao, N.; Yudha, S.; Nogami, T.; Yamamoto, Y. *Angew. Chem., Int. Ed.* **2005**, *44*, 5526. (f) Yanada, R.; Obika, S.; Kono, H.; Takemoto, Y. *Angew. Chem., Int. Ed.* **2006**, *45*, 3822.
- For metal-mediated azomethine ylide formation/dipolar cycloaddition, see: (a) Padwa, A.; Dean, D. C.; Zhi, L. *J. Am. Chem. Soc.* **1992**, *114*, 593. For recent reviews on azomethine ylide cycloaddition, see: (b) Najera, C.; Sansano, J. M. *Curr. Org. Chem.* **2003**, *7*, 1105. (c) Coldham, I.; Hufton, R. *Chem. Rev.* **2005**, *105*, 2765. For cycloadditions of metal-containing azomethine ylides from alkynyl imines, see: (d) Kusama, H.; Takaya, J.; Iwasawa, N. *J. Am. Chem. Soc.* **2002**, *124*, 11592. (e) Takaya, J.; Kusama, H.; Iwasawa, N. *Chem. Lett.* **2004**, *33*, 16.
- Beeler, A. B.; Su, S.; Singleton, C. A.; Porco, J. A., Jr. *J. Am. Chem. Soc.* **2007**, *129*, 1413.
- See Supporting Information for complete experimental details.
- Asao, N.; Iso, K.; Yudha, S. *Org. Lett.* **2006**, *8*, 4149.
- See for example: Suga, H.; Inoue, K.; Inoue, S.; Kakehi, A. *J. Am. Chem. Soc.* **2002**, *124*, 14836.
- Matsumoto, K.; Uchida, T.; Konishi, H.; Watanabe, Y.; Aoyama, K.; Asahi, M. *Chem. Lett.* **1987**, 807.
- For dihydropyrroles resembling compound **41**, see: Bacu, E.; Samson-Belei, D.; Nowogrocki, G.; Couture, A.; Grandclaudon, P. *Org. Biomol. Chem.* **2003**, *1*, 2377.
- (a) Cuadro, A. M.; Alvarez-Builla, J.; Vaquero, J. J. *Heterocycles* **1989**, *29*, 57. (b) Gandasegui, M. T.; Alvarez-Builla, J.; Florencio, F. *Heterocycles* **1994**, *37*, 174.

JA072737V